





7. Tracez les composés suivants.

a. triméthyle - 2,3,5 heptane

b. éthyle - 4 diméthyle - 2,4 octane

c. Butyle - 5 triméthyle - 2,4,5 éthyle - 3 nonane

d. diméthyle - 3,5 propyle - 7 décane

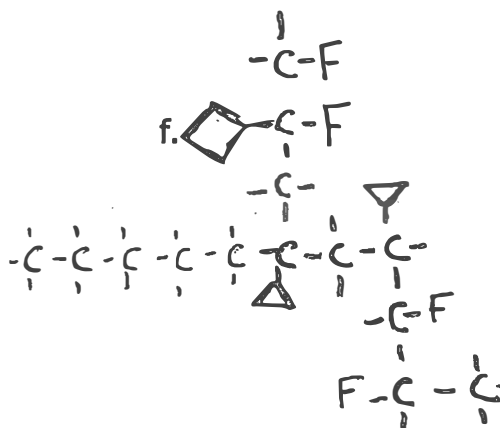
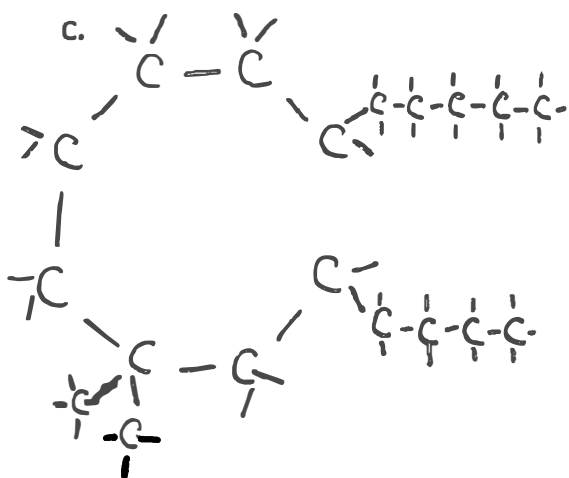
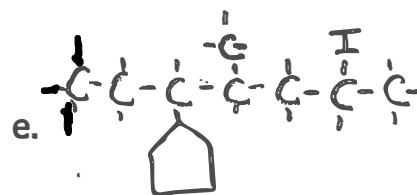
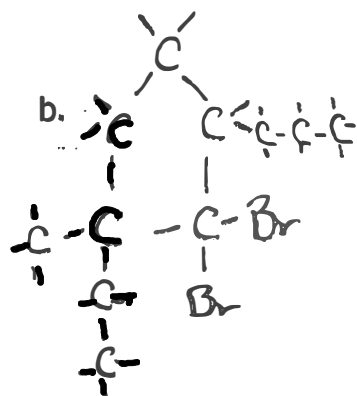
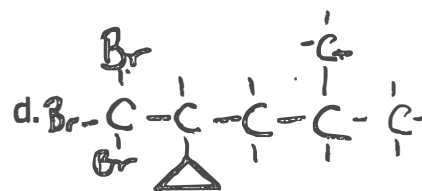
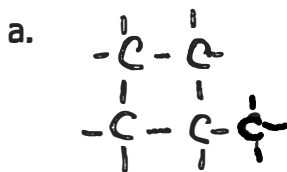
e. tétraméthyle - 2,3,5,7 pentyle - 3 heptane



# La chimie organique

## Les cyclanes et les isomères

1. Nommez les composés suivants.



2. Qu'est-ce que c'est un isomère? \_\_\_\_\_

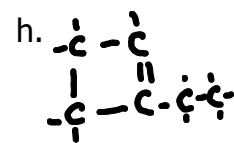
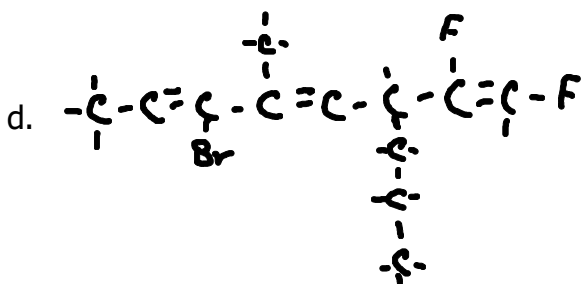
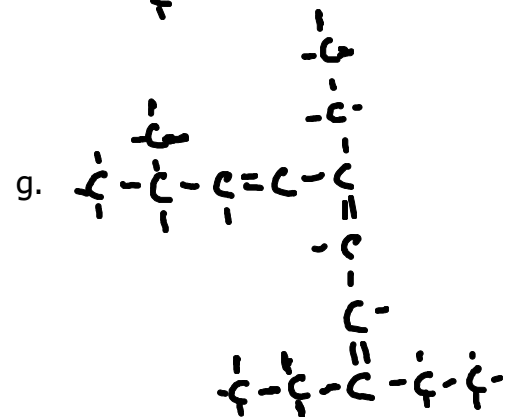
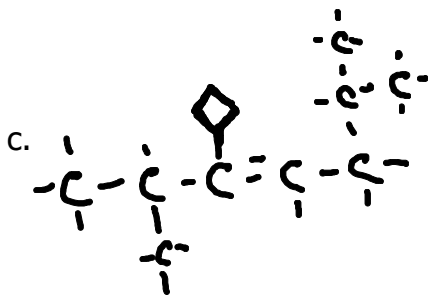
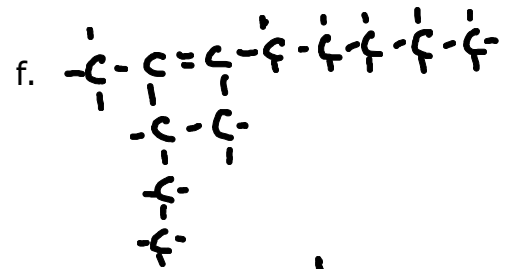
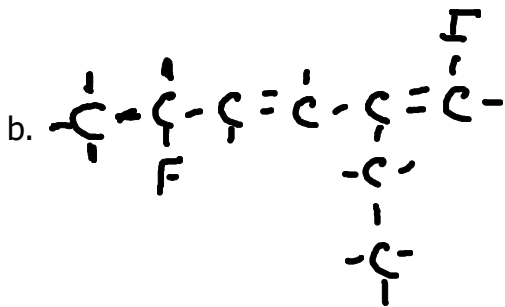
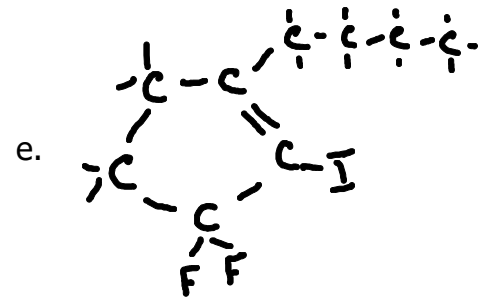
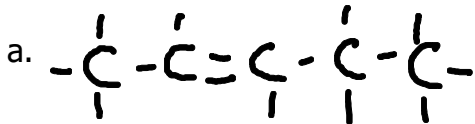
3. Tracez et nommez tous les isomères de heptane.

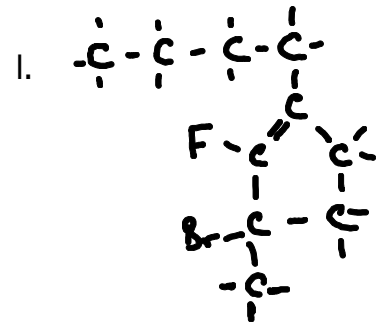
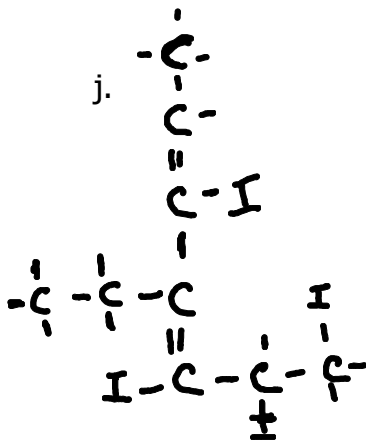
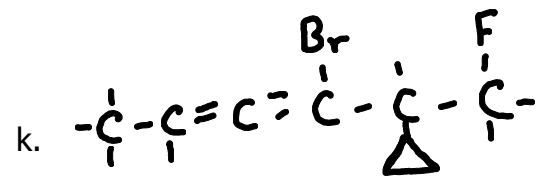
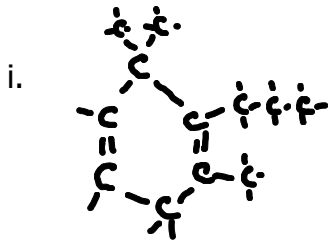
4. Tracez et nommez 8 de les isomères de nonane

# La chimie organique

## Les Alcènes

1. Quelle est la formule générale des alcènes?
2. Pourquoi est-ce qu'on dit que les alcènes sont non-saturés?
3. Nommez les composés suivants.





4. Tracez les composés suivants.

a. dibromo-1,3 fluoro-4 éthyle-5 diméthyle-2,7 octadiène-1,4

b. tétraiodo-2,4,5,5 butyle-3 propyle-4 cyclohexène-1



c. Éthyle-2 hexyle-4 diméthyle-3,3 heptadiène-1,5

d. Tracez et nommez un alcène avec 1 double liaison, deux halogènes, et deux radicaux d'hydrocarbures.

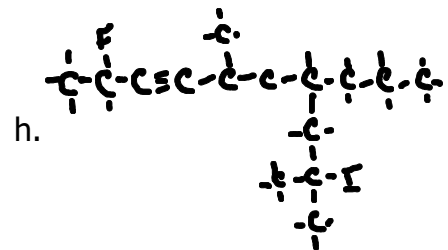
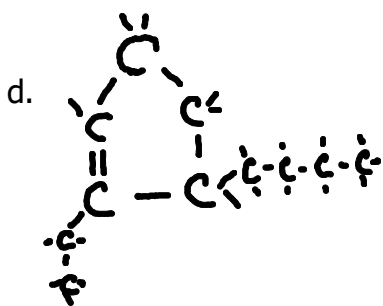
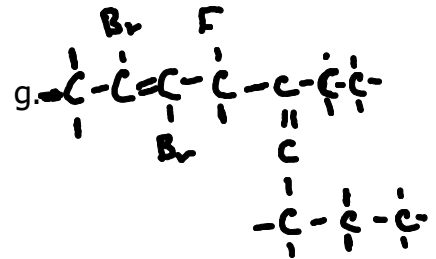
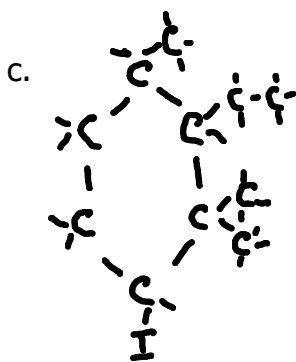
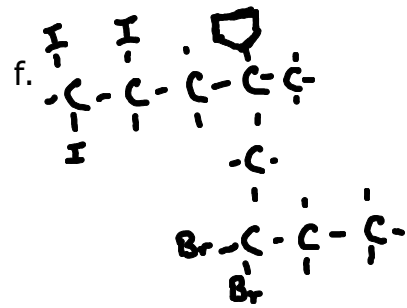
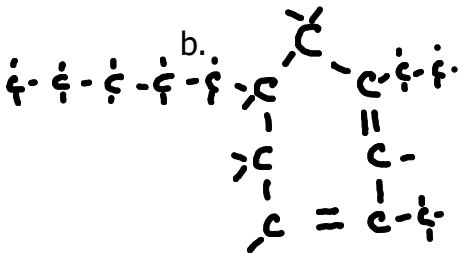
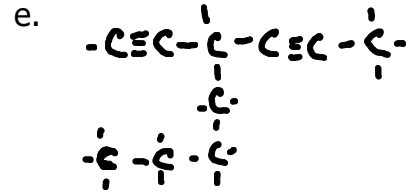
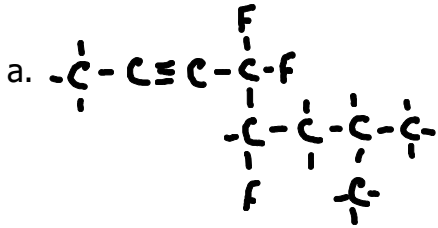
e. Tracez et nommez un cycloalcène avec 2 liaisons doubles et 3 radicaux d'hydrocarbures.

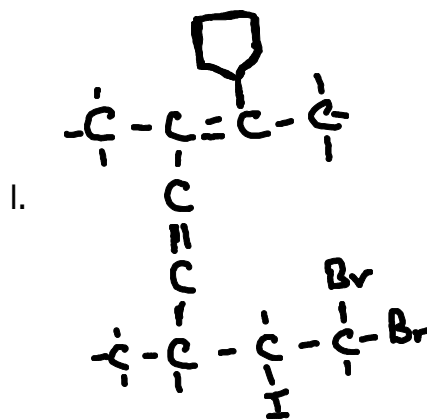
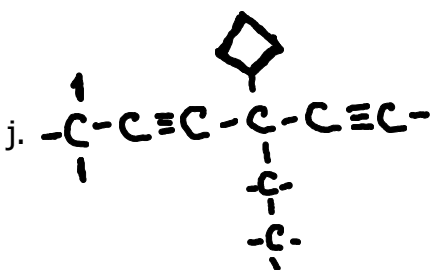
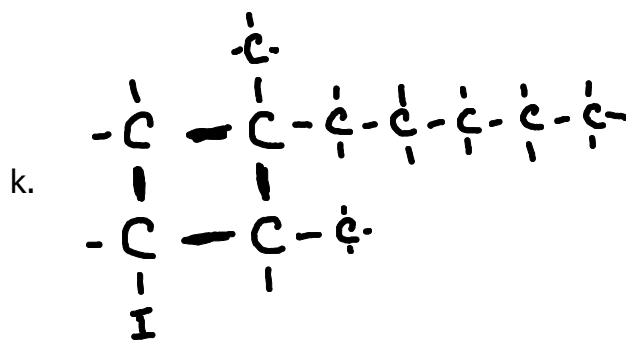
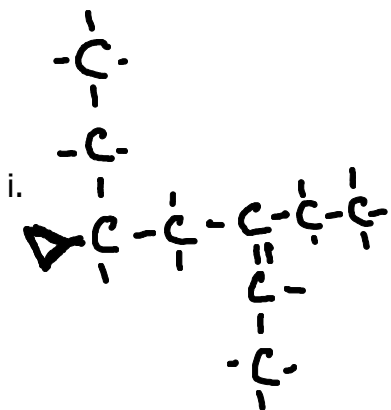


# Les Alcynes

1. Quelle est la formule générale des alcynes?

2. Nommez les composés suivants. (mélange d'alcane, alcène et alcyne)





3. Tracez les composés suivants.

a. trifluoro-1,1,4 diméthyle-4,5 éthyle-7  
nonadiyne-2,6

b. cyclobutyle-2 propyle-4 cyclohexadiène-1,3